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# Power law statistics of rippled graphene nanoflakes

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**Abstract** Graphene nanoflakes (GNFs) are predicted to possess novel magnetic, optical, and spintronic properties. They have recently been synthesized and a number of applications are being studied. Here we investigate the statistical properties of rippled GNFs (50–5,000 atoms) at T = 300 K. An adjacency matrix is calculated from the coordinates and we find that the free energy, enthalpy, entropy, and atomic displacement all show power law behavior. The vibrational energy versus the Wiener index also shows power law character. We distinguish between using Euclidean topographical indices and compare them to topological ones. These properties are determined from atomic coordinates using MATLAB routines.

Keywords Graphene nanoflakes · Statistical mechanics · Wiener index · MATLAB

## **1** Introduction

Graphene [1] is a 2D allotrope of carbon [diamond, graphite, fullerene, carbon nanotubes, (CNTs)] that has been found to have exceptional electronic and mechanical properties. Recently, scientists have started considering limiting the size of the 2D sheet form of graphene, thus defining graphene nanoribbons, GNRs, and graphene nanoflakes, GNFs. One experimental method of attempting this is by using catalytic metal nanoparticles [2] cutting along crystallographic planes. When restricting the 2D sheet character of graphene by two dimensions in the plane, one creates GNFs, which theory predicts to have unique magnetic, optical, and spintronic properties.

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GNFs may have a range of magnetic character, from ferromagnetic [3,4], to ferrimagnetic [5,6], to antiferromagnetic [7], depending on GNF geometry and topological frustration. These properties have enabled the design of spintronic NOR and NAND gates [7], which in principle can operate at room temperature. A spin-valve type effect [8] has been investigated in triangular GNFs. Additionally, hydrogenation can change the magnetic and electronic [9] character of the GNF. The optical properties of GNFs span the entire visible spectrum [10], opening the route to new nano optical devices. The electronic band structure as a function of increasing size [11] of triangular GNFs shows that semiconducting behavior exists for small GNFs.

An interesting question to consider when thinking about the thermal character of GNFs, is whether they are actually stable, or would they transform to another form of carbon, such as the fullerene or CNT shapes? This question has been modeled by density functional theory (DFT) and *ab initio* molecular dynamics (MD) calculations, with the result [12,13] that GNFs do not transform to a different allotrope, but neither are they truly 2D, in the sense that the structure of the GNF becomes buckled and rippled at elevated temperatures. Annealing of the structures [12] found out-of-plane distortions as the temperature increased to 2,400 K, but no fundamental change in structure occurred.

Experimentally, the progress in creating GNFs is behind that of theoretical modeling. GNFs have been created using a 'top-down' approach from exfoliation of graphite [14], to chemical vapor deposition [15, 16], to arc-discharged material [17]. Among the properties examined, electron field emission has predominated [14, 16]. To date, a true nanoengineering 'bottom-up' approach remains open to development, although some progress [18, 19] has been made, especially for GNRs [20]. Thus, the truly exotic properties of GNFs, such as those already mentioned, remain to be explored.

#### 2 Methods

Here we represent GNFs as graphs G = (V, E) where nodes represent atoms and edges represent physical interactions between those atoms. We create a graph of the GNF by creating bonds (links) between nearest neighbors up to 1.3 times the shortest neighbor distance from rippled MD models of graphene [21]. These vary from 50 to 5,000 atoms and as created, are the asymmetric form [12] of a GNF. The entries of the adjacency matrix are given by

$$A = \begin{cases} H(r_c - r_{ij}) \ i \neq j \\ 0 \qquad i = j \end{cases}$$
(1)

where the Heaviside step function  $H(r_c - r_{ij}) = 1$  if  $r_c < 1.3*$  (shortest distance to  $r_{ij}$ ), and *i* and *j* represent atomic sites, and  $r_c$  is the cutoff value. We also use the actual Euclidean distances in the adjacency matrix [22,23], so that  $H(r_c - r_{ij}) = e_{ij}$ , the Euclidean distance between atoms.

We use a standard approach to modeling the entropy, free energy, and enthalpy as discussed in the literature [24,25]. These can all be determined from the appropriate adjacency matrix. The Wiener index [26], is calculated as



Fig. 1 MATLAB plot of a 3D rippled GNF with 500 atoms

$$W = \frac{1}{2} \sum_{i=1}^{n} \sum_{j=1}^{n} d_{ij}$$
(2)

where *n* is the number of atoms and  $d_{ij}$  is the shortest *path* distance between atoms *i* and *j* in the molecular graph. Most commonly, the distances between atoms  $d_{ij} = 1$ , and in the Euclidean form [22],  $d_{ij} = e_{ij}$ , so that we calculate W<sub>E</sub>, the Euclidean 3D Wiener index. Atomic coordinates are used to generate the adjacency matrix and the statistical calculations proceed from a MATLAB routine.

#### **3 Results**

In Fig. 1, we show a plot of a rippled GNF with 500 atoms. In our modeling, the edge and corner atoms exist as shown in the figure, and as tabulated for the 50–5,000 atom structures in Table 1. The coordination number of the GNF, N<sub>C</sub>, increases from 2.44 to 2.9404. The bond length varies around 1.42 Å. A histogram of the various bond lengths in the 500–5,000 atom GNFs is shown in Fig. 2, and the bond lengths range from 1.33 to 1.52 Å.

The adjacency matrix allows us to calculate the statistical mechanics parameters in a standard way [23,24]. The partition function is

$$Z(G,\beta) = \operatorname{Tr}(e^{\beta \mathbf{A}}) \tag{3}$$

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N	N <sub>1</sub>	N <sub>2</sub>	N <sub>3</sub>	N <sub>C</sub>	Ave bond Å
50	5	18	27	2 44	1 4248
100	7	26	67	2.60	1.4219
500	16	62	422	2.812	1.4228
1,000	22	86	892	2.87	1.4229
2,500	35	138	2,327	2.9168	1.4239
5,000	50	198	4,752	2.9404	1.4238

**Table 1** N is the number of atoms in the GNF,  $N_1$ ,  $N_2$ , and  $N_3$ , are the number of one, two, and threefold coordinated atoms in the GNF, and  $N_C$  is the coordination number

The average bond length is in Angstroms



Fig. 2 Histogram of the bond lengths (in Angstroms) of the 500-5,000 GNFs

where **A** is the adjacency matrix for the graph G, and  $\beta = 1/(k_B T)$ . At T = 300 K, we have  $\beta = 38.68173/eV$ . The entropy can be determined as

$$S(G,\beta) = -k_B \sum_j \lambda_j p_j + k_B \ln(Z) \sum_j p_j \tag{4}$$

where  $\lambda_i$  is an eigenvalue of **A** and

$$p_j = \frac{e^{\beta\lambda_j}}{Z(G,\beta)} \tag{5}$$

is the probability that the ensemble occupies a microstate j. The free energy is the natural logarithm of the partition function,

$$F(G,\beta) = -\frac{\ln Z(G,\beta)}{\beta}$$
(6)

and the enthalpy can be defined as follows

$$H(G,\beta) = -\frac{1}{Z(G,\beta)} \operatorname{Tr}(\mathbf{A}e^{\beta \mathbf{A}}).$$
(7)

We then plot the free energy, enthalpy, and entropy, per bond, versus the number of bonds in the GNF. This results in plots with good power law [27] regression statistics as shown in Fig. 3. The distinction between Fig. 3a, b is that in (a), we have used an adjacency matrix with Euclidean distances, and in (b), we have used the standard adjacency matrix with zeros and ones. The best-fit equations in (a) have different leading coefficients, so that the entropy and enthalpy coincide (neglecting the sign difference) for small (50 atoms) GNFs, and since the slope is different, the plots diverge for larger GNFs. Since we use a value of the Boltzmann constant in terms of eV, the values of our thermodynamic calculations for the state of N<sub>B</sub> = 1 and N = 2 are: entropy, 1.3272 eV/K, enthalpy, -2.6674 eV, and free energy, -0.0883 eV. These quantities are divided by the number of bonds in the GNF and plotted versus N<sub>B</sub>, to give a power law plot. The asymptote of zero for large N<sub>B</sub> makes intuitive sense, since if we imagine the data/bond is finite, then as the number of bonds becomes large, we have zero as a limit. Note that the free energy and enthalpy have their signs reversed to allow them to be plotted.

We now proceed to calculate some related parameters of interest. In the harmonic approximation [28], the frequency of the stretching vibration of a carbon-carbon bond is given by

$$\nu = \frac{1}{2\pi} \sqrt{\frac{k}{\mu}} \tag{8}$$

where k is the force constant (about 305 N/m for graphene [29]) between nearest neighbor carbon atoms and  $\mu$  is the reduced mass. The total intramolecular energy,  $E_{int}$  can be divided into two parts; a variable term  $U_{var}$  and a constant term  $U_{con}$ . We conclude that the variable term can be written [28] as

$$U \cong \sum \frac{1}{\sqrt{M_1 M_2/M}} \tag{9}$$

where the proportionality constant depends on the force constant, but not the mass dependence. If we rewrite this in terms of our GNF, we have

$$U = \sum \sqrt{\frac{n}{Cn_1n_2}} \tag{10}$$

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Fig. 3 Plots of the free energy, enthalpy, and entropy/bond versus the number of bonds. The data clearly exhibits power law character. **a** This plot uses a Euclidean adjacency matrix. **b** This plot uses a standard adjacency matrix



Fig. 4 Vibrational energy, U, versus the standard Wiener index, W, and the 3D Euclidean Wiener index  $W_E$ . The data follows a power law

where  $n = n_1 + n_2$ , and  $n_1$  and  $n_2$  are the number of carbon atoms on the two sides of the vibrating bond. We may compare this to the relationship [26,28] for the Wiener index

$$W = \sum n_1 n_2. \tag{11}$$

Now in order to linearize U, we note that the minimum and maximum values of the product  $n_1n_2$  are (n - 1) and  $(n/2)^2$ , respectively. The middle value is

$$\xi = \frac{n^2}{8} + \frac{n-1}{2} \tag{12}$$

and we expand U as a power function about  $x = \xi$ 

$$(Cx)^{-1/2} \approx f(\xi) + f'(\xi)(x - \xi)$$
 (13)

where

$$f(\xi) = \frac{1}{\sqrt{C}} \sqrt{\frac{8n}{n^2 + 4n - 4}}.$$
(14)

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Fig. 5 Power law plots of  $N_B$ , U, W, and  $\langle XX \rangle$  versus the number of carbon atoms, N. The statistics and regression of the data are quite good

If we write C = 12 amu, then to a first approximation we have

$$U \approx \frac{1}{\sqrt{12}} N_B \sqrt{\frac{8n}{n^2 + 4n - 4}} \tag{15}$$

Since *n* is the number of atoms in the GNF and we calculate  $N_B$  through each iteration of the MATLAB program, we can determine the approximate values for U. Now both U and W have power law character, so we plot U versus W and  $W_E$  in Fig. 4. These results show good regression features. The only distinction is that  $W_E$  has slightly larger values, but behaves in the same manner as W. The MATLAB code for W has been examined previously [30], but we have adapted it to work with an adjacency matrix.

In the harmonic approximation, we may also consider the vibrational excitation energy from the static position of the GNF. Previous calculations [31], show that the summed displacement may be calculated as

$$\langle X_i X_i \rangle = \sqrt{\sum_{i=1}^n (\Delta x_i)^2} = \sqrt{\frac{W}{nk\beta}}$$
(16)

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where  $\beta = 38.68173/eV$ , *k* is the force constant between carbon atoms (305 N/m), *n* is the number of atoms, and W is the Wiener index. In Fig. 5, we show the dependence of N<sub>B</sub>, U, W, and  $\langle XX \rangle$  on the number of carbon atoms N. The values of the summed displacement range from 0.4760 to 14.7106 Å for 50–5,000 atoms, respectively. If we were to consider the average displacement per bond, the values would be 0.0039 and 0.000747 Å, for 50 and 5,000 atoms, respectively. The data all show good regression features.

### **4** Conclusion

We have determined the power law behavior of the free energy, enthalpy, entropy, and atomic displacement of GNFs consisting of 50–5,000 atoms at T = 300 K. The vibrational energy versus the Wiener index also shows power law character. There are some minor distinctions when using the Euclidean Wiener index, which we have included for completeness.

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